

Physics

- The frequencies for series limit of Balmer and Paschen series respectively are ' ν_1 ' and ' ν_3 '. If frequency of first line of Balmer series is ' ν_2 ' then the relation between ' ν_1 ', ' ν_2 ' and ' ν_3 ' is
 - $\nu_1 - \nu_2 = \nu_3$
 - $\nu_1 + \nu_3 = \nu_2$
 - $\nu_1 + \nu_2 = \nu_3$
 - $\nu_1 - \nu_3 = 2\nu_1$
- When three capacitors of equal capacities are connected in parallel and one of the same capacity is connected in series with its combination. The resultant capacity is $3.75 \mu\text{F}$. The capacity of each capacitor is
 - $5 \mu\text{F}$
 - $6 \mu\text{F}$
 - $7 \mu\text{F}$
 - $8 \mu\text{F}$
- Sensitivity of moving coil galvanometer is ' S '. If a shunt of $\left(\frac{1}{8}\right)$ th of the resistance of galvanometer is connected to moving coil galvanometer, its sensitivity becomes
 - $\frac{S}{3}$
 - $\frac{S}{6}$
 - $\frac{S}{9}$
 - $\frac{S}{12}$
- Two unknown resistances are connected in two gaps of a meter-bridge. The null point is obtained at 40cm from left end. A 30Ω resistance is connected in series with the smaller of the two resistances, the null point shifts by 20cm to the right end. The value of smaller resistance in Ω is
 - 12
 - 24
 - 36
 - 48

5. In Fraunhofer diffraction pattern, slit width is 0.2 mm and screen is at 2m away from the lens. If wavelength of light used is 5000\AA , then the distance between the first minimum on either side of the central maximum is
(θ is small and measured in radian)
- a. 10^{-1}m b. 10^{-2}m
c. $2 \times 10^{-2}\text{m}$ d. $2 \times 10^{-1}\text{m}$
6. In series LCR circuit $R = 18\ \Omega$ and impedance is $33\ \Omega$. An rms voltage 220V is applied across the circuit. The true power consumed in AC circuit is
- a. 220 W b. 400 W
c. 600 W d. 800 W
7. Two parallel plate air capacitors of same capacity C are connected in series to a battery of emf E . Then one of the capacitors is completely filled with dielectric material of constant K . The change in the effective capacity of the series combination is
- a. $\frac{C}{2} \left[\frac{K-1}{K+1} \right]$ b. $\frac{2}{C} \left[\frac{K-1}{K+1} \right]$
c. $\frac{C}{2} \left[\frac{K+1}{K-1} \right]$ d. $\frac{C}{2} \left[\frac{K-1}{K+1} \right]^2$
8. The polarising angle for transparent medium is ' θ ' and ' v ' is the speed of light in that medium. Then the relation between ' θ ' and ' v ' is
(c = velocity of light in air)
- a. $\theta = \tan^{-1} \left(\frac{v}{c} \right)$ b. $\theta = \cot^{-1} \left(\frac{v}{c} \right)$
c. $\theta = \sin^{-1} \left(\frac{v}{c} \right)$ d. $\theta = \cos^{-1} \left(\frac{v}{c} \right)$
9. Two identical light waves having phase difference ' ϕ ' propagate in same direction. When they superpose, the intensity of resultant wave is proportional to
- a. $\cos^2 \phi$ b. $\cos^2 \frac{\phi}{2}$
c. $\cos^2 \frac{\phi}{3}$ d. $\cos^2 \frac{\phi}{4}$
10. For a transistor, α_{dc} and β_{dc} are the current ratios, then the value of $\frac{\beta_{dc} - \delta_{dc}}{\alpha_{dc} \cdot \beta_{dc}}$
- a. 1 b. 1.5
c. 2 d. 2.5
11. A radioactive element has rate of disintegration 10,000 disintegrations per minute at a particular instant. After four minutes it becomes 2500 disintegrations per minute. The decay constant per minute is
- a. $0.2 \log_e 2$ b. $0.5 \log_e 2$
c. $0.6 \log_e 2$ d. $0.8 \log_e 2$
12. When the same monochromatic ray of light travels through glass slab and through water, the number of waves in glass slab of thickness 6cm is same as in water column of height 7cm. If refractive index of glass is 1.5, then refractive index of water is
- a. 1.258 b. 1.269
c. 1.286 d. 1.310
13. If the electron in hydrogen atom jumps from second Bohr orbit to ground state and difference between energies of the two states is radiated in the form of photons. If the work function of the material is 4.2 eV, then stopping potential is
[Energy of electron in n th orbit = $-\frac{13.6}{n^2}$ eV]
- a. 2V b. 4V c. 6V d. 8V
14. The magnetic moment of electron due to orbital motion is proportional to
(n = principal quantum numbers)
- a. $\frac{1}{n^2}$ b. $\frac{1}{n}$ c. n^2 d. n
15. Photodiode is a device
- a. which is always operated in reverse bias
b. which is always operated in forward bias
c. in which photo current is independent of intensity of incident radiation
d. which may be operated in forward or reverse bias.

16. A wheel of moment of inertia 2 kg m^2 is rotating about an axis passing through centre and perpendicular to its plane at a speed 60 rad/s . Due to friction, it comes to rest in 5 minutes. The angular momentum of the wheel three minutes before it stops rotating is

- a. $24 \text{ kg m}^2/\text{s}$
 b. $48 \text{ kg m}^2/\text{s}$
 c. $72 \text{ kg m}^2/\text{s}$
 d. $96 \text{ kg m}^2/\text{s}$

17. The equation of the progressive wave is

$$y = 3 \sin \left[\pi \left(\frac{t}{3} - \frac{x}{5} \right) + \frac{\pi}{4} \right], \text{ where } x \text{ and } y$$

are in metre and time in second. Which of the following is correct?

- a. Velocity $v = 1.5 \text{ m/s}$
 b. Amplitude $A = 3 \text{ cm}$
 c. Frequency $f = 0.2 \text{ Hz}$
 d. Wavelength $\lambda = 10 \text{ m}$

18. Two spherical black bodies have radii ' r_1 ' and ' r_2 '. Their surface temperature are ' T_1 ' and ' T_2 '. If they radiate same power, then

$$\frac{r_2}{r_1} \text{ is}$$

- a. $\frac{T_1}{T_2}$ b. $\frac{T_2}{T_1}$
 c. $\left(\frac{T_1}{T_2} \right)^2$ d. $\left(\frac{T_2}{T_1} \right)^2$

19. The closed and open organ pipes have same length. When they are vibrating simultaneously in first overtone, produce three beats. The length of open pipe is made $\frac{1}{3}$ rd and closed pipe is made three

times the original, the number of beats produced will be

- a. 8 b. 14
 c. 17 d. 20

20. A lift of mass ' m ' is connected to a rope which is moving upward with maximum acceleration ' a '. For maximum safe stress, the elastic limit of the rope is ' T '. The minimum diameter of the rope is ($g = \text{gravitational acceleration}$)

- a. $\left[\frac{2m(g+a)}{\pi T} \right]^{\frac{1}{2}}$ b. $\left[\frac{4m(g+a)}{\pi T} \right]^{\frac{1}{2}}$
 c. $\left[\frac{m(g+a)}{\pi T} \right]^{\frac{1}{2}}$ d. $\left[\frac{m(g+a)}{2\pi T} \right]^{\frac{1}{2}}$

21. A solid sphere of mass 2 kg is rolling on a frictionless horizontal surface with velocity 6 m/s . It collides on the free end of an ideal spring whose other end is fixed. The maximum compression produced in the spring will be

(Force constant of the spring = 36 N/m)

- a. $\sqrt{14} \text{ m}$ b. $\sqrt{2.8} \text{ m}$ c. $\sqrt{14} \text{ m}$ d. $\sqrt{0.7} \text{ m}$

22. A flywheel at rest is to reach an angular velocity of 24 rad/s in 8 second with constant angular acceleration. The total angle turned through during this interval is

- a. 24 rad b. 48 rad
 c. 72 rad d. 96 rad

23. Two uniform wires of the same material are vibrating under the same tension. If the first overtone of the first wire is equal to the second overtone of the second wire and radius of the first wire is twice the radius of the second wire, then the ratio of the lengths of the first wire to second wire is

- a. $\frac{1}{3}$ b. $\frac{1}{4}$ c. $\frac{1}{5}$ d. $\frac{1}{6}$

24. When one end of the capillary is dipped in water, the height of water column is ' h '.

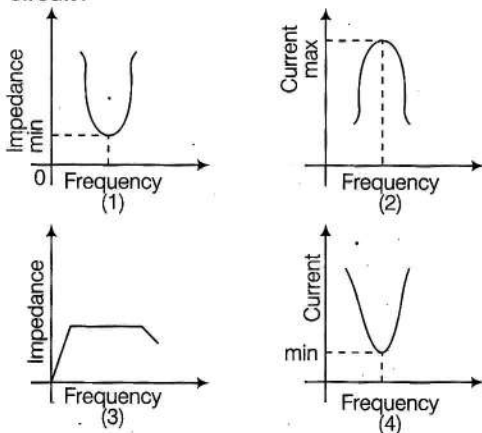
The upward force of 105 dyne due to surface tension is balanced by the force due to the weight of water column. The inner circumference of the capillary is (Surface tension of water = $7 \times 10^{-2} \text{ N/m}$)

- a. 1.5 cm b. 2 cm c. 2.5 cm d. 3 cm

25. For a rigid diatomic molecule, universal gas constant $R = nC_p$, where ' C_p ' is the molar specific heat at constant pressure and ' n ' is a number. Hence n is equal to
 a. 0.2257 b. 0.4 c. 0.2857 d. 0.3557
26. An ideal gas has pressure ' p ', volume ' V ' and absolute temperature ' T '. If ' m ' is the mass of each molecules and ' K ' is the Boltzmann constant, then density of the gas is
 a. $\frac{pm}{MT}$ b. $\frac{KT}{pm}$ c. $\frac{Km}{pT}$ d. $\frac{pK}{Tm}$
27. A big water drop is formed by the combination of ' n ' small water drops of equal radii. The ratio of the surface energy of ' n ' drops to the surface energy of big drop is
 a. $n^2:1$ b. $n:1$ c. $\sqrt{n}:1$ d. $\sqrt[3]{n}:1$
28. The ratio of binding energy of a satellite at rest on earth's surface to the binding energy of a satellite of same mass revolving around the earth at a height h above the earth's surface is (R = radius of the earth).
 a. $\frac{2(R+h)}{R}$ b. $\frac{R+h}{2}$
 c. $\frac{R+h}{R}$ d. $\frac{R}{R+h}$
29. A particle performing SHM starts equilibrium position and its time period is 16 seconds. After 2 seconds its velocity is π m/s. Amplitude of oscillation is $\left(\cos 45^\circ = \frac{1}{\sqrt{2}}\right)$
 a. $2\sqrt{2}$ m b. $4\sqrt{2}$ m c. $6\sqrt{2}$ m d. $8\sqrt{2}$ m
30. In sonometer experiment, the string of length ' L ' under tension vibrates in second overtone between two bridges. The amplitude of vibration is maximum at
 a. $\frac{L}{3}, \frac{2L}{3}, \frac{5L}{6}$ b. $\frac{L}{8}, \frac{L}{4}, \frac{L}{2}$
 c. $\frac{L}{2}, \frac{L}{4}, \frac{L}{6}$ d. $\frac{L}{6}, \frac{L}{2}, \frac{5L}{6}$
31. The depth ' d ' at which the value of acceleration due to gravity becomes $\frac{1}{n}$ times the value at the earth's surface is (R = radius of earth)
 a. $d = R\left(\frac{n}{n-1}\right)$ b. $d = R\left(\frac{n-1}{2n}\right)$
 c. $d = R\left(\frac{n-1}{n}\right)$ d. $d = R^2\left(\frac{n-1}{n}\right)$
32. A particle is performing SHM starting extreme position. Graphical representation shows that, between-displacement and acceleration, there is a phase difference of
 a. 0 rad b. $\frac{\pi}{4}$ rad
 c. $\frac{\pi}{2}$ rad d. π rad
33. The fundamental frequency of an air column in a pipe closed at one end is 100 Hz. If the same pipe is open at both the ends, the frequencies produced in Hz are
 a. 100, 200, 300, 400, ...
 b. 100, 300, 500, 700, ...
 c. 200, 300, 400, 500, ...
 d. 200, 400, 600, 800
34. For a particle moving in vertical circle, the total energy at different positions along the path
 a. is conserved
 b. increases
 c. decreases
 d. may increase or decrease
35. A simple pendulum of length ' L ' has mass ' M ' and it oscillates freely with amplitude ' A '. At extreme position, its potential energy is (g = acceleration due to gravity)
 a. $\frac{MgA^2}{2L}$ b. $\frac{MgA}{2L}$
 c. $\frac{MgA^2}{L}$ d. $\frac{2MgA^2}{L}$

36. On a photosensitive material, when frequency of incident radiation is increased by 30%, kinetic energy of emitted photoelectrons increases from 0.4 eV to 0.9 eV. The work function of the surface is
 a. 1 eV b. 1.267 eV c. 1.4 eV d. 1.8 eV

37. Out of the following graphs, which graphs shows the correct relation (graphical representation) for LC parallel resonant circuit?



- a. 1 b. 2 c. 3 d. 4

38. According to de-Broglie hypothesis, the wavelength associated with moving electron of mass ' m ' is ' λ_e '. Using mass energy relation and Planck's quantum theory, the wavelength associated with photon is ' λ_p '. If the energy (E) of electron and photon is same, then relation between ' λ_e ' and ' λ_p ' is

- a. $\lambda_p \propto \lambda_e$ b. $\lambda_p \propto \lambda_e^2$
 c. $\lambda_p \propto \sqrt{\lambda_e}$ d. $\lambda_p \propto \frac{1}{\lambda_e}$

39. A parallel plate air capacity ' C ' farad, potential ' V ' volt and energy ' E ' joule. When the gap between the plates is completely filled with dielectric
 a. both V and E increase
 b. both V and E decrease
 c. V decrease, E increases
 d. V increases, E decrease

40. The resistivity of potentiometer wire is 40×10^{-8} ohm-metre and its area of cross-section is $8 \times 10^{-6} \text{m}^2$. If 0.2 ampere current is flowing through the wire, then the potential gradient of the wire is
 a. 10^{-1}V/m b. 10^{-2}V/m
 c. 10^{-3}V/m d. 10^{-4}V/m

41. A ceiling fan rotates about its own axis with some angular velocity. When the fan is switched off, the angular velocity becomes $\left(\frac{1}{4}\right)$ th of the original in time ' t ' and ' n ' revolutions are made in that time. The number of revolutions made by the fan during the time interval between switch off and rest are (Angular retardation is uniform)

- a. $\frac{4n}{15}$ b. $\frac{8n}{15}$ c. $\frac{16n}{15}$ d. $\frac{32n}{15}$

42. A disc of moment of inertia ' I_1 ' is rotating in horizontal plane about an axis passing through a centre and perpendicular to its plane with constant angular speed ' ω_1 '. Another disc of moment of inertia ' I_2 ' having zero angular speed is placed co-axially on a rotating disc. Now, both the discs are rotating with constant angular speed ' ω_2 '. The energy lost by the initial rotating disc is

- a. $\frac{1}{2} \left[\frac{I_1 + I_2}{I_1 I_2} \right] \omega_1^2$ b. $\frac{1}{2} \left[\frac{I_1 I_2}{I_1 - I_2} \right] \omega_1^2$
 c. $\frac{1}{2} \left[\frac{I_1 - I_2}{I_1 I_2} \right] \omega_1^2$ d. $\frac{1}{2} \left[\frac{I_1 I_2}{I_1 + I_2} \right] \omega_1^2$

43. A particle performs linear SHM at a particular instant, velocity of the particle is ' u ' and acceleration is ' α ' while at another instant velocity is ' v ' and acceleration is ' β ' ($0 < \alpha < \beta$). The distance between the two position is

- a. $\frac{u^2 - v^2}{\alpha + \beta}$ b. $\frac{u^2 + v^2}{\alpha + \beta}$
 c. $\frac{u^2 - v^2}{\alpha - \beta}$ d. $\frac{u^2 + v^2}{\alpha - \beta}$

44. The observer is moving with velocity v_0 towards the stationary source of sound and then after crossing moves away from the source with velocity v_0 . Assume that the medium through which the sound waves travel is at rest. If v is the velocity of sound and n is the frequency emitted by the source, then the difference between apparent frequencies heard by the observer is
- a. $\frac{2nv_0}{v}$ b. $\frac{nv_0}{v}$ c. $\frac{v}{2nv_0}$ d. $\frac{v}{nv_0}$
45. A metal rod of length ' L ' and cross-sectional area ' A ' is heated through ' $T^\circ\text{C}$ '. What is the force required to prevent the expansion of the rod lengthwise?
- a. $\frac{YA\alpha T}{(1-\alpha T)}$ b. $\frac{YA\alpha T}{(1+\alpha T)}$
 c. $\frac{(1-\alpha T)}{YA\alpha T}$ d. $\frac{(1+\alpha T)}{YA\alpha T}$
46. Two coils P and Q are kept near each other. When no current flows through coil P and current increases in coil Q at the rate 10A/s , the emf in coil P is 15mV . When coil Q carries no current and current of 1.8A flows through coil P , the magnetic flux linked with the coil Q is
- a. 1.4 m Wb b. 2.2 mWb
 c. 2.7 m Wb d. 2.9 m Wb
47. In Young's double experiment, in air interference pattern second minimum is observed exactly in front of one slit. The distance between the two coherent source is ' d ' and the distance between source and screen is ' D '. The wavelength of light source used is
- a. $\frac{d^2}{D}$ b. $\frac{d^2}{2D}$ c. $\frac{d^2}{3D}$ d. $\frac{d^2}{4D}$
48. In communication system, the process of superimposing a low frequency signal on a high frequency wave is known as
- a. repeater b. attenuation
 c. modulation d. demodulation
49. A bar magnet has length 3 cm , cross-sectional area 2 cm^2 and magnetic moment 3 Am^2 . The intensity of magnetisation of bar magnet is
- a. $2 \times 10^5\text{ A/m}$ b. $3 \times 10^5\text{ A/m}$
 c. $4 \times 10^5\text{ A/m}$ d. $5 \times 10^5\text{ A/m}$
50. The magnetic flux near the axis and inside the air core solenoid of length 60 cm carrying current ' I ' is $157 \times 10^{-6}\text{ Wb}$. Its magnetic moment will be (cross-sectional area of a solenoid is very small as compared to its length, $\mu_0 = 4\pi \times 10^{-7}\text{ SI unit}$)
- a. 0.25 A b. 0.50 A
 c. 0.75 A d. 1 A

Chemistry

51. The work done during combustion of $9 \times 10^{-2}\text{ kg}$ of ethane, $\text{C}_2\text{H}_6(g)$ at 300 K is (Given $R = 8.314\text{ J deg}^{-1}\text{ mol}^{-1}$, atomic mass $\text{C} = 12, \text{H} = 1$)
- a. 6.236 kJ b. -6.236 kJ
 c. 18.71 kJ d. -18.71 kJ
52. What type of sugar molecule is present in DNA?
- a. D-3-deoxyribose b. D-ribose
 c. D-2-deoxyribose d. D-glucopyranose
53. The molality of solution containing 15.20 g of urea, (molar mass = 60) dissolved in 150 g of water is
- a. 1.689 mol kg^{-1} b. $0.1689\text{ mol kg}^{-1}$
 c. $0.5922\text{ mol kg}^{-1}$ d. $0.2533\text{ mol kg}^{-1}$
54. The acid, which contains both $-\text{OH}$ and $-\text{COOH}$ groups is
- a. phthalic acid
 b. adipic acid
 c. glutaric acid
 d. salicylic acid

55. Identify the compound, in which phosphorus exists in the oxidation state of +1.
- Phosphonic acid (H_3PO_3)
 - Phosphinic acid (H_3PO_2)
 - Pyrophosphorus acid ($\text{H}_4\text{P}_2\text{O}_5$)
 - Orthophosphoric acid (H_3PO_4)
56. Identify the weakest oxidising agent among the following.
- Li^+
 - Na^+
 - Cd^{2+}
 - I_2
57. The monomers used in preparation of dextran are
- lactic acid and glycollic acid
 - 3-hydroxy butanoic acid and 3-hydroxy pentanoic acid
 - styrene and 1, 3-butadiene
 - hexamethylenediamine and adipic acid
58. Which among the following compounds does not act as reducing agent?
- H_2O
 - H_2S
 - H_2Se
 - H_2Te
59. Which of the following processes is not used to preserve the food?
- Irradiation
 - Addition of salts
 - Addition of heat
 - Hydration
60. In case of substituted aniline the group which decreases the basic strength is
- $-\text{OCH}_3$
 - $-\text{CH}_3$
 - $-\text{NH}_2$
 - $-\text{C}_6\text{H}_5$
61. (+2) 2-methylbutan-1-ol (-) 2-methylbutan-1-ol have different values for which property?
- Boiling point
 - Relative density
 - Refractive index
 - Specific rotation
62. Which among the following is not a mineral of iron?
- Haematite
 - Magnetite
 - Magnesite
 - Siderite
63. Nitration of which among the following compounds yields cyclohexite?
- Formaldehyde
 - Benzaldehyde
 - Urotropine
 - Acetaldehyde ammonia
64. Calculate the work done during compression of 2 mol of an ideal gas from a volume of 1 m^3 to 10 dm^3 at 300K against a pressure of 100 KPa.
- 99 kJ
 - + 99 kJ
 - + 22.98 kJ
 - 22.98 kJ
65. Which element among the following does form $p\pi - p\pi$ multiple bonds?
- Arsenic
 - Nitrogen
 - Phosphorus
 - Antimony
66. Which of the following statement(s) is/are incorrect in case of Hofmann bromamide degradation?
- Reaction is useful for decreasing length of carbon chain by one carbon atom
 - It gives tertiary amine
 - It gives primary amine
 - Aqueous or alco. KOH is used with bromine
67. Which of the following statement(s) is/are incorrect for pair of elements Zr-Hf?
- Both possess same number of valence electrons
 - Both have identical sizes
 - Both have almost identical radii
 - Both of these belong to same period of periodic table
68. Aldehyde or ketones when treated with $\text{C}_6\text{H}_5\text{NH-NH}_2$, the product formed is
- semicarbazone
 - phenylhydrazone
 - hydrazone
 - oxime
69. Solubility of which among the following solids in water changes slightly with temperature?
- KNO_3
 - NaNO_3
 - KBr
 - NaBr

70. What is the quantity of hydrogen gas liberated when 46 g sodium reacts with excess ethanol?
(Given atomic mass of Na = 23)
- a. 2.4×10^{-3} kg b. 2.0×10^{-3} kg
c. 4.0×10^{-3} kg d. 2.4×10^{-2} kg
71. *Tert*-butyl methyl ether on treatment with hydrogen iodide in cold gives
- a. *tert*-butyl iodide and methyl iodide
b. *tert*-butyl alcohol and methyl alcohol
c. *tert*-butyl alcohol and methyl iodide
d. *tert*-butyl iodide and methyl alcohol
72. Name the process that is employed to refine aluminium.
- a. Hall's process b. Mond process
c. Hoop's process d. Serperck's process
73. The colour and magnetic nature of manganate ion (MnO_4^{2-}) is
- a. green, paramagnetic
b. purple, diamagnetic
c. green, diamagnetic
d. purple, paramagnetic
74. The osmotic pressure of solution containing 34.2 g of cane sugar (molar mass = 342 g mol^{-1}) in 1L of solution at 20°C is (Given $R = 0.082 \text{ L atm K}^{-1} \text{ mol}^{-1}$)
- a. 2.40 atm b. 3.6 atm
c. 24 atm d. 0.0024 atm
75. In assigning *R-S* configuration, which among the following groups has highest priority?
- a. $-\text{SO}_3\text{H}$ b. $-\text{COOH}$
c. $-\text{CHO}$ d. $-\text{C}_6\text{H}_5$
76. Which of the following is used as antiseptic?
- a. Chloramphenicol
b. Bithional
c. Cimetidine
d. Chlordiazepoxide
77. In preparation of sulphuric acid from sulphur dioxide in lead chamber process. What substance is used as a catalyst?
- a. Manganese dioxide
b. Vanadium pentoxide
c. Nitric oxide
d. Rauey nickel
78. The correct charge on and co-ordination number of 'Fe' in $\text{K}_3[\text{Fe}(\text{CN})_6]$ is
- a. + 2, 4 b. + 3, 6
c. + 2, 6 d. + 3, 3
79. Which among the following reactions is an example of pseudo first order reaction?
- a. Inversion of cane sugar
b. Decomposition of H_2O_2
c. Conversion of cyclopropane to propene
d. Decomposition of N_2O_5
80. The amine, which reacts with *p*-toluenesulphonyl chloride to give a clear solution, which on acidification gives insoluble compound is
- a. $\text{C}_2\text{H}_5\text{NH}_2$ b. $(\text{C}_2\text{H}_5)_2\text{NH}$
c. $(\text{C}_2\text{H}_5)_3\text{N}$ d. $\text{CH}_3\text{NHC}_2\text{H}_5$
81. Which among the following equation represents Arrhenius equation?
- a. $k = A_e^{E_a/RT}$ b. $k = A \cdot e^{RT/E_a}$
c. $k = \frac{A}{e^{E_a/RT}}$ d. $k = \frac{A}{e^{RT/E_a}}$
82. Which of the following compound will give positive iodoform test?
- a. Isopropyl alcohol
b. Propionaldehyde
c. Ethylphenyl ketone
d. Benzyl alcohol
83. The first law of thermodynamics for isothermal process is
- a. $q = -W$ b. $\Delta U = W$
c. $\Delta U = q_v$ d. $\Delta U = -q_v$

84. The conversion of ethyl bromide using sodium iodide and dry acetone, this reaction is known as
 a. Swarts reaction
 b. Finkelstein reaction
 c. Sandmeyer reaction
 d. Stephen reaction
85. What is the hybridisation of carbon atoms in fullerene?
 a. sp^3 b. sp c. sp^2 d. dsp^3
86. What is the SI unit of conductivity?
 a. Sm b. Sm^{-1} c. Sm^2 d. Sm^{-2}
87. Which of the following is Baeyer's reagent?
 a. Alkaline $KMnO_4$ b. Acidic $K_2Cr_2O_7$
 c. Alkaline $Na_2Cr_2O_7$ d. MnO_2
88. What is the chief constituent of pyrex glass?
 a. B_2O_3 b. SiO_2 c. Al_2O_3 d. Na_2O
89. Which of the following compounds has lowest boiling point?
 a. *n*-butyl alcohol b. Iso-butyl alcohol
 c. *Tert*-butyl alcohol d. *Sec*-butyl alcohol
90. Identify the invalid equation
 a. $\Delta H = \sum H_{\text{products}} - \sum H_{\text{reactants}}$
 b. $\Delta H = \Delta U + p\Delta V$
 c. $\Delta H^{\circ}_{\text{(reaction)}} = \sum H^{\circ}_{\text{(product bonds)}} - \sum H^{\circ}_{\text{(reactant bonds)}}$
 d. $\Delta H = \Delta U + \Delta nRT$
91. The rate constant for a first order reaction is $7.0 \times 10^{-4} \text{ s}^{-1}$. If initial concentration of reactant is 0.080 M, what is the half life of reaction?
 a. 990 s b. 79.2 s
 c. 12375 s d. $10.10 \times 10^{-4} \text{ s}$
92. The polymer used in making handles of cookers and frying pans is
 a. bakelite b. nylon-2-nylon-6
 c. orlon d. polyvinyl chloride
93. Which halogen has the highest value of negative electron gain enthalpy?
 a. Fluorine b. Chlorine
 c. Bromine d. Iodine
94. What is the actual volume occupied by water molecules present in 20 cm^3 of water?
 a. 20 cm^3 b. 10 cm^3
 c. 40 cm^3 d. 24.89 cm^3
95. Which of the following co-ordinate complexes is an exception to EAN rule? (Given atomic number Pt = 78, Fe = 26, Zn = 30, Cu = 29)
 a. $[Pt(NH_3)_6]^{4+}$ b. $[Fe(CN)_6]^{4-}$
 c. $[Zn(NH_3)_4]^{2+}$ d. $[Cu(NH_3)_4]^{2+}$
96. Which among the following equations represents the reduction reaction taking place in lead accumulator at positive electrode, while it is being used as a source of electrical energy?
 a. $Pb \rightarrow Pb^{2+}$ b. $Pb^{4+} \rightarrow Pb$
 c. $Pb^{2+} \rightarrow Pb$ d. $Pb^{4+} \rightarrow Pb^{2+}$
97. For which among the following equimolar aqueous solutions Van't Hoff factor has the lowest value?
 a. Aluminium chloride
 b. Potassium sulphate
 c. Ammonium chloride
 d. Urea
98. The amino acid, which is basic in nature is
 a. histidine b. tyrosine
 c. proline d. valine
99. Which element among the following does not form diatomic molecules?
 a. Argon b. Oxygen
 c. Nitrogen d. Bromine
100. A molecule of stachyose contains how many carbon atoms?
 a. 6 b. 12
 c. 18 d. 24

Mathematics

1. The number of principal solutions of $\tan 2\theta = 1$ is

- a. one b. two
c. three d. four

2. The objective function $z = 4x_1 + 5x_2$, subject to $2x_1 + x_2 \geq 7$, $2x_1 + 3x_2 \leq 15$, $x_2 \leq 3$, $x_1, x_2 \geq 0$ has minimum value at the point

- a. on X-axis
b. on Y-axis
c. at the origin
d. on the line parallel to X-axis

3. If z_1 and z_2 are z-coordinates of the points of trisection of the segment joining the points $A(2, 1, 4)$, $B(-1, 3, 6)$, then $z_1 + z_2 =$

- a. 1 b. 4
c. 5 d. 10

4. The maximum value of

$$f(x) = \frac{\log x}{x} \quad (x \neq 0, x \neq 1) \text{ is}$$

- a. e b. $\frac{1}{e}$
c. e^2 d. $\frac{1}{e^2}$

5. $\int_0^1 x \tan^{-1} x \, dx =$

- a. $\frac{\pi}{4} + \frac{1}{2}$ b. $\frac{\pi}{4} - \frac{1}{2}$
c. $\frac{1}{2} - \frac{\pi}{4}$ d. $-\frac{\pi}{4} - \frac{1}{2}$

6. The statement pattern $(\sim p \wedge q)$ is logically equivalent to

- a. $(p \vee q) \vee \sim p$ b. $(p \vee q) \wedge \sim p$
c. $(p \wedge q) \rightarrow p$ d. $(p \vee q) \rightarrow p$

7. If $g(x)$ is the inverse function of $f(x)$ and

$$f'(x) = \frac{1}{1+x^4}, \text{ then } g'(x) \text{ is}$$

- a. $1 + [g(x)]^4$ b. $1 - [g(x)]^4$
c. $1 + [f(x)]^4$ d. $\frac{1}{1 + [f(x)]^4}$

8. The inverse of the matrix $\begin{bmatrix} 1 & 0 & 0 \\ 3 & 3 & 0 \\ 5 & 2 & -1 \end{bmatrix}$ is

a. $-\frac{1}{3} \begin{bmatrix} -3 & 0 & 0 \\ 3 & 1 & 0 \\ 9 & 2 & -3 \end{bmatrix}$ b. $-\frac{1}{3} \begin{bmatrix} -3 & 0 & 0 \\ 3 & -1 & 0 \\ -9 & -2 & 3 \end{bmatrix}$

c. $-\frac{1}{3} \begin{bmatrix} 3 & 0 & 0 \\ 3 & -1 & 0 \\ -9 & -2 & 3 \end{bmatrix}$ d. $-\frac{1}{3} \begin{bmatrix} -3 & 0 & 0 \\ -3 & -1 & 0 \\ -9 & -2 & 3 \end{bmatrix}$

9. If $\int \frac{1}{\sqrt{9-16x^2}} dx = \alpha \sin^{-1}(\beta x) + c$, then

$$\alpha + \frac{1}{\beta} =$$

- a. 1 b. $\frac{7}{12}$
c. $\frac{19}{12}$ d. $\frac{9}{12}$

10. $O(0, 0)$, $A(1, 2)$, $B(3, 4)$ are the vertices of ΔOAB . The joint equation of the altitude and median drawn from O is

- a. $x^2 + 7xy - y^2 = 0$
b. $x^2 + 7xy + y^2 = 0$
c. $3x^2 - xy - 2y^2 = 0$
d. $3x^2 + xy - 2y^2 = 0$

11. If the function $f(x) = \left[\tan\left(\frac{\pi}{4} + x\right) \right]^{\frac{1}{x}}$ for $x \neq 0$

is $= K$ for $x = 0$

continuous at $x = 0$, then $K = ?$

- a. e b. e^{-1}
c. e^2 d. e^{-2}

12. For an invertible matrix A if

$$A(\text{adj}A) = \begin{bmatrix} 10 & 0 \\ 0 & 10 \end{bmatrix}, \text{ then } |A| =$$

- a. 100
b. -100
c. 10
d. -10

13. The solution of the differential equation

$$\frac{dy}{dx} = \tan\left(\frac{y}{x}\right) + \frac{y}{x} \text{ is}$$

- a. $\cos\left(\frac{y}{x}\right) = cx$ b. $\sin\left(\frac{y}{x}\right) = cx$
 c. $\cos\left(\frac{y}{x}\right) = cy$ d. $\sin\left(\frac{y}{x}\right) = cy$

14. In ΔABC , if $\sin^2 A + \sin^2 B = \sin^2 C$ and $l(AB) = 10$, then the maximum value of the area of ΔABC is

- a. 50 b. $10\sqrt{2}$
 c. 25 d. $25\sqrt{2}$

15. If $x = f(t)$ and $y = g(t)$ are differentiable

functions of t , then $\frac{d^2y}{dx^2}$ is

- a. $\frac{f'(t) \cdot g''(t) - g'(t) \cdot f''(t)}{[f'(t)]^3}$
 b. $\frac{f'(t) \cdot g''(t) - g'(t) \cdot f''(t)}{[f'(t)]^2}$
 c. $\frac{g'(t) \cdot f''(t) - f'(t) \cdot g''(t)}{[f'(t)]^3}$
 d. $\frac{g'(t) \cdot f''(t) + f'(t) \cdot g''(t)}{[f'(t)]^3}$

16. The equation of line equally inclined to coordinate axes and passing through $(-3, 2, -5)$ is

- a. $\frac{x+3}{1} = \frac{y-2}{1} = \frac{z+5}{1}$
 b. $\frac{x+3}{-1} = \frac{y-2}{1} = \frac{z+5}{-1}$
 c. $\frac{x+3}{-1} = \frac{y-2}{1} = \frac{z+5}{1}$
 d. $\frac{x+3}{-1} = \frac{z-5}{1} = \frac{y-2}{-1}$

17. If $\int_0^{\frac{\pi}{2}} \log \cos x \, dx = \frac{\pi}{2} \log\left(\frac{1}{2}\right)$, then

$$\int_0^{\frac{\pi}{2}} \log \sec x \, dx =$$

- a. $\frac{\pi}{2} \log\left(\frac{1}{2}\right)$ b. $1 - \frac{\pi}{2} \log\left(\frac{1}{2}\right)$
 c. $1 + \frac{\pi}{2} \log\left(\frac{1}{2}\right)$ d. $\frac{\pi}{2} \log 2$

18. A boy tosses fair coin 3 times. If he gets $2X$ for X heads, then his expected gain equals to

- a. 1 b. $\frac{3}{2}$ c. 3 d. 4

19. Which of the following statement pattern is a tautology?

- a. $p \vee (q \rightarrow p)$
 b. $\sim q \rightarrow \sim p$
 c. $(q \rightarrow p) \vee (\sim p \leftrightarrow q)$
 d. $p \wedge \sim p$

20. If the angle between the planes

$$r \cdot (m\hat{i} - \hat{j} + 2\hat{k}) + 3 = 0 \text{ and}$$

$$r \cdot (2\hat{i} - m\hat{j} - \hat{k}) - 5 = 0 \text{ is } \frac{\pi}{3}, \text{ then } m =$$

- a. 2 b. ± 3
 c. 3 d. -2

21. If the origin and the points

$P(2, 3, 4)$, $Q(1, 2, 3)$ and $R(x, y, z)$ are coplanar, then

- a. $x - 2y - z = 0$ b. $x + 2y + z = 0$
 c. $x - 2y + z = 0$ d. $2x - 2y + z = 0$

22. If lines represented by equation $px^2 - qy^2 = 0$ are distinct, then

- a. $pq > 0$ b. $pq < 0$
 c. $pq = 0$ d. $p + q = 0$

23. Let $\square PQRS$ be a quadrilateral. If M and N are the mid-points of the sides PQ and RS respectively, then $\overline{PS} + \overline{QR} =$

- a. $3 \overline{MN}$ b. $4 \overline{MN}$
 c. $2 \overline{MN}$ d. $2 \overline{NM}$

24. If slopes of lines represented by $kx^2 + 5xy + y^2 = 0$ differ by 1, then $k =$

- a. 2 b. 3 c. 6 d. 8

25. If vector r with dc's l, m, n is equally inclined to the coordinate axes, then the total number of such vectors is

- a. 4 b. 6 c. 8 d. 2

26. If $\int \frac{1}{(x^2 + 4)(x^2 + 9)} dx = A \tan^{-1} \frac{x}{2} + B$

$\tan^{-1} \left(\frac{x}{3} \right) + C$, then $A - B =$

- a. $\frac{1}{6}$ b. $\frac{1}{30}$
 c. $-\frac{1}{30}$ d. $-\frac{1}{6}$

27. If α and β are roots of the equation

$x^2 + 5|x| - 6 = 0$, then the value of $|\tan^{-1} \alpha - \tan^{-1} \beta|$ is

- a. $\frac{\pi}{2}$ b. 0
 c. π d. $\frac{\pi}{4}$

28. If $x = a \left(t - \frac{1}{t} \right)$, $y = a \left(t + \frac{1}{t} \right)$, where t is the parameter, then $\frac{dy}{dx} = ?$

- a. $\frac{y}{x}$ b. $-\frac{x}{y}$
 c. $\frac{x}{y}$ d. $-\frac{y}{x}$

29. The point on the curve $y = \sqrt{x-1}$, where the tangent is perpendicular to the line $2x + y - 5 = 0$ is

- a. (2, -1) b. (10, 3)
 c. (2, 1) d. (5, -2)

30. If $\int \sqrt{\frac{x-5}{x-7}} dx = A \sqrt{x^2 - 12x + 35}$

$+ \log|x - 6 + \sqrt{x^2 - 12x + 35}| + C$, then $A =$

- a. -1 b. $\frac{1}{2}$ c. $-\frac{1}{2}$ d. 1

31. At random variable $X \sim B(n, p)$, if values of mean and variance of X are 18 and 12 respectively, then total number of possible values of X are

- a. 54 b. 55
 c. 12 d. 18

32. The area of the region bounded by the lines $y = 2x + 1$, $y = 3x + 1$ and $x = 4$ is

- a. 16 sq unit b. $\frac{121}{3}$ sq unit
 c. $\frac{121}{6}$ sq unit d. 8 sq unit

33. A box contains 6 pens, 2 of which are defective. Two pens are taken randomly from the box. If r.v. X : number of defective pens obtained, then standard deviation of $X =$

- a. $\pm \frac{4}{3\sqrt{5}}$ b. $\frac{8}{3}$
 c. $\frac{16}{45}$ d. $\frac{4}{3\sqrt{5}}$

34. If the volume of spherical ball is increasing at the rate of $4\pi \text{ cm}^3/\text{s}$, then the rate of change of its surface area when the volume is $288\pi \text{ cm}^3$, is

- a. $\frac{4}{3}\pi \text{ cm}^2/\text{s}$ b. $\frac{2}{3}\pi \text{ cm}^2/\text{s}$
 c. $4\pi \text{ cm}^2/\text{s}$ d. $2\pi \text{ cm}^2/\text{s}$

35. If $f(x) = \log(\sec^2 x)^{\cot^2 x}$ for $x \neq 0$
 $= K$ for $x = 0$

is continuous at $x = 0$, then K is

- a. e^{-1} b. 1
 c. e d. 0

36. If c denotes the contradiction, then dual of the compound statement $\sim p \wedge (q \vee c)$ is

- a. $\sim p \vee (q \wedge t)$ b. $\sim p \wedge (q \vee t)$
 c. $p \vee (\sim q \vee t)$ d. $\sim p \vee (q \wedge c)$

37. The differential equation of all parabolas whose axis is Y-axis, is

- a. $x \frac{d^2y}{dx^2} - \frac{dy}{dx} = 0$ b. $x \frac{d^2y}{dx^2} + \frac{dy}{dx} = 0$
 c. $\frac{d^2y}{dx^2} - y = 0$ d. $\frac{d^2y}{dx^2} - \frac{dy}{dx} = 0$

38. $\int_0^3 [x] dx = \dots\dots\dots$, where $[x]$ is greatest integer function.

- a. 3 b. 0
 c. 2 d. 1

39. The objective function of LPP defined over the convex set attains its optimum value at

- a. atleast two of the corner points
- b. all the corner points
- c. atleast one of the corner points
- d. None of the corner points

40. If the inverse of the matrix $\begin{bmatrix} \alpha & 14 & -1 \\ 2 & 3 & 1 \\ 6 & 2 & 3 \end{bmatrix}$

does not exist, then the value of α is

- a. 1
- b. -1
- c. 0
- d. -2

41. If $f(x) = x$ for $x \leq 0$

$= 0$ for $x > 0$, then $f(x)$ at $x = 0$ is

- a. continuous but not differentiable
- b. not continuous but differentiable
- c. continuous and differentiable
- d. not continuous and not differentiable

42. The equation of the plane through $(-1, 1, 2)$, whose normal makes equal acute angles with coordinate axes is

- a. $\mathbf{r} \cdot (\hat{i} + \hat{j} + \hat{k}) = 2$
- b. $\mathbf{r} \cdot (\hat{i} + \hat{j} + \hat{k}) = 6$
- c. $\mathbf{r} \cdot (3\hat{i} - 3\hat{j} + 3\hat{k}) = 2$
- d. $\mathbf{r} \cdot (\hat{i} - \hat{j} + \hat{k}) = 3$

43. Probability that a person will develop immunity after vaccinations is 0.8. If 8 people are given the vaccine, then probability that all develop immunity is =

- a. $(0.2)^8$
- b. $(0.8)^8$
- c. 1
- d. ${}^8C_6 (0.2)^6 (0.8)^2$

44. If the distance of points $2\hat{i} + 3\hat{j} + \lambda\hat{k}$ from the plane $\mathbf{r} \cdot (3\hat{i} + 2\hat{j} + 6\hat{k}) = 13$ is 5 units, then $\lambda =$

- a. $6, -\frac{17}{3}$
- b. $6, \frac{17}{3}$
- c. $-6, -\frac{17}{3}$
- d. $-6, \frac{17}{3}$

45. The value of $\cos^{-1}\left(\cot\left(\frac{\pi}{2}\right)\right) + \cos^{-1}\left(\sin\left(\frac{2\pi}{3}\right)\right)$ is

- a. $\frac{2\pi}{3}$
- b. $\frac{\pi}{3}$
- c. $\frac{\pi}{2}$
- d. π

46. The particular solution of the differential equation $x dy + 2y dx = 0$, when $x = 2$, $y = 1$ is

- a. $xy = 4$
- b. $x^2y = 4$
- c. $xy^2 = 4$
- d. $x^2y^2 = 4$

47. ΔABC has vertices at

$A \equiv (2, 3, 5)$, $B \equiv (-1, 3, 2)$ and $C \equiv (\lambda, 5, \mu)$. If the median through A is equally inclined to the axes, then the values of λ and μ respectively are

- a. 10, 7
- b. 9, 10
- c. 7, 9
- d. 7, 10

48. For the following distribution function $F(x)$ of a rv. X .

x	1	2	3	4	5	6
$F(x)$	0.2	0.37	0.48	0.62	0.85	1

$P(3 < x \leq 5) =$

- a. 0.48
- b. 0.37
- c. 0.27
- d. 1.47

49. The lines $\frac{x-1}{2} = \frac{y+1}{2} = \frac{z-1}{4}$ and

$\frac{x-3}{1} = \frac{y-k}{2} = \frac{z}{1}$ intersect each other at point

- a. $(-2, -4, 5)$
- b. $(-2, -4, -5)$
- c. $(2, 4, -5)$
- d. $(2, -4, -5)$

50. $\int \frac{\sec^8 x}{\operatorname{cosec} x} dx =$

- a. $\frac{\sec^8 x}{8} + c$
- b. $\frac{\sec^7 x}{7} + c$
- c. $\frac{\sec^6 x}{6} + c$
- d. $\frac{\sec^9 x}{9} + c$